

## SODIUM HYPOCHLORITE

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Rev.0

### 1. Product and Company Identification

Product Name	Sodium hypochlorite
Structure Formula	NaOCl
CAS Number	7681-52-9
Synonyms	Bleach, HICHLOR
Manufacturer's Name	AGC Chemicals (Thailand) Co., Ltd.
Address	
Head Office	24 <sup>th</sup> floor, Bangkok Insurance Building, 25 South Sathorn Road, Bangkok 10120, Thailand Tel. (662) 679-1600 Fax. (662) 677-3177
Phrapadaeng Factory	202 Suksawasdi Rd., Km.17, Samutprakarn, 10290, Thailand Tel. (662) 463-6345-8 Fax. (662) 463-3728
Rayong Factory	4 Soi G-12, Pakorn Songkrohrad Rd., Eastern Industrial Estate, Map-Ta-Phut, Muang, Rayong 21150, Thailand Tel. (038) 683-572-5 Fax. (038) 683-576

### 2. Composition / Information on Ingredient

Substance	Concentration (by weight)
Sodium hypochlorite	≥ 10% AB.Cl <sub>2</sub>

### 3. Physical / Chemical Properties

Molecular Weight	74.4	pH	11
Melting Point (°C)	-6 (5% solution)	Density (g/cm <sup>3</sup> )	1.2
Boiling Point (°C)	Decomposition above 40°C	Status	Liquid
Appearance and Odor	Greenish yellow with a chlorine odor		
Solubility in water	Soluble in all proportions.		

### 4. Fire and Explosion Hazard Data

Extinguishing Media	Use any means suitable for extinguishing surrounding fire and/or materials.
Fire Hazard Comments	Fire-exposure or excessive heat may cause the rupture of containers.
Special Fire Fighting Procedures	Use water to cool the containers.
Protective Equipments for Fire Fighters	Wear fire-resistant suit, chemical resistant suit and self-contained breathing apparatus.
Flash Point (°C)	Non-flammable
Autoignition Temperature (°C)	Non-flammable

### NFPA Symbol



Flammability	0	: Will not burn
Reactivity	2	: May undergo violent change at elevated temperature and pressure.
Health	2	: Intense or continued exposure could cause temporary incapacitation or possible residual injury unless prompt medical treatment is given.
Special data	OXY	: Oxidizer

## 5. Reactivity Data

Stabilization	Solution decomposes slowly. Rate of decomposition depends on heat and light.
Prevention	Keep away from heat and light.
Explosion Data	Not explosive
Reaction with Water	Will not occur
Oxidation	Oxidizer
Hazardous Decomposition Products	Chlorine, oxygen, sodium chlorate
Protection from Decomposition Products	Use chemical cartridge respirator containing the chlorine cartridge.

## 6. Health Hazard Data

**Special Precaution** A corrosive chemical.

### Health Effects

**Routes of Entry** Inhalation, skin, eye and swallow

**Hazard (Skin, Eye and Mucous membrane)**

Causes irritation.

### Effects of Short-Term (Acute) Exposure

**Inhalation** : May irritate nose and the respiratory tract.

**Skin Contact** : May cause skin irritation. In severe cases, burn may occur.

**Eye Contact** : May cause severe eye irritation.

**Ingestion** : May cause irritation and pain. Causes severe burn to mouth and stomach, vomiting, shock and death.

### Effects of Long-Term (Chronic) Exposure

**Skin** : Causes dryness, cracking and dermatitis.

## First Aid Procedure

<b>Skin Contact</b>	Remove contaminated clothing and shoes under running water for at least 15 minutes. Obtain medical attention immediately.
<b>Eye Contact</b>	Flush with running water for at least 15 minutes, occasionally lifting the eyelids. Do not allow the contaminated water into the non-affected eye. Obtain medical attention immediately.
<b>Inhalation</b>	Move victim to fresh air. If breathing is difficult, give oxygen. Obtain medical attention immediately.
<b>Ingestion</b>	Never give anything by mouth if victim is unconscious. Rinse mouth thoroughly with water. Do not induce vomiting. Drink 240 to 300 ml. of water. Obtain medical attention immediately.
<b>Exposure Guidelines</b>	TLV-C: 2 mg/m <sup>3</sup>

## Toxicological Information

### Acute Toxicity

<b>LD<sub>50</sub> ingestion (mg/l)</b>	5800	(Rat)
<b>LD<sub>50</sub> skin (mg/l)</b>	> 10000	(Rabbit)
<b>LC<sub>50</sub> inhalation (ppm)</b>	> 10000 for 1 h.	(Rat)

**Eye Contact** Severe irritation

**Skin Contact** Irritation

**Sub-Acute Toxicity** Slightly liver damage on rat when administered with 0.4% of sodium hypochloride for 13 consecutive weeks.

**Allergenic Effects** Not information available

### Chronic Toxicity

**Carcinogenic Effects** Not classified as carcinogen.

**Embryologic Effects** Not information available

**Teratogenic Effects** Not information available

**Mutagenic Effects** Not information available

**Neurogenic Effects** Not information available



## 7. Precaution for Handling and Use

### Handling

#### Warning

Avoid generating mist for decreasing the dispersion. Do not allow react with acid which lead to chlorine gas.

#### Precaution

All equipments will be cleaned before using.

#### Ventilation

Adequate ventilation should be provided.

#### Safety Handling

Use smallest possible amount in designated areas with adequate ventilation. Prepare appropriate safety measures and protective equipment. Keep containers tightly closed.

#### Incompatible Materials

Reducing agents, Strong acids, Nitrogen compound, Copper, Nickel, Cobalt

### Exposure Controls

#### Personal Protection

Restrict access to exposure area. Use appropriate personal protective equipments. Have a well-ventilated system.

#### Environment Protection

Prevent liquid run-off into sewers, which lead to water ways. Use sand or soil to make a dike.

#### Spill and Leakage Procedures

Contain spill with soil, sand, or absorbent material. Sweep up material and place into a suitable labelled disposal container. Flush area with water.

### Waste Disposal Method

#### Products

React with reducing agents such as sodium metabisulfite and then neutralize with sodium carbonate or sodium thiosulfate.

#### Empty Containers

Clean up with water and neutralize with sodium carbonate. Dispose containers with all compliance in law regulations.

## 8. Control Measure

### Engineering Controls

Totally enclose processes and personal. Control the condition of process. Normal ventilation is generally adequate. If generated heat or vapors, local exhaust ventilation should be provided.

### Respiratory Protection

Not specification but chemical cartridge respirator with a chlorine cartridge should be provided.

### Body Protection

Protective clothing

### Hand Protection

Chemical resistant gloves


### Eye Protection

Chemical safety goggles, or glasses. Face shield may be used in properly.

### Others Protection

Chemical resistant boots, Eyewash fountain and safety shower. Do not eat, drink or smoke in work areas.

## 9. Regulatory Information

ORANGE SYMBOL	LABEL
<div data-bbox="129 1668 317 1805"> <div>80</div> <div>1791</div> </div> <div data-bbox="336 1668 671 1787"> <div>: Corrosive substance and react violently with water.</div> <div>: UN Number</div> </div>	<div data-bbox="831 1641 1007 1823">  </div> <div data-bbox="1031 1659 1501 1794"> <div>For transportation.</div> <div>Label sizing : more than 250 x 250 mm.</div> <div>Picture sizing : 12.5 mm. far from edge 5 mm.</div> </div>

### Hazchem Code

2R
1791

2 : Use water spray or fog to reduce or direct vapors.  
 R : Use chemical protective full body and self contained breathing apparatus.  
 Dilute with water before release to sewers, water ways.  
 1791 : UN number

## 10. Transportation Information

UN Number	1791	UN Class	8
UN Packing Group	III	IMDG-Ems Number	8-08
IMDG-Class	8	IMDG Packing Group	III
IATA-Class	8	Tank Number	L4BV (+)
IATA-Packing Group	III		

## 11. Other Informations

Bioaccumulation	Not available
Ecotoxicological Information	Not available